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What are all the metalloids on the periodic table

Metalloids are elements from the periodic table with properties that lie between metals and non-metals. The following ScienceStruck article will cover some information related to metalloids. The first person to come up with a periodic table of elements was Dmitri Ivanovich Mendeleev, a Russian chemist. He came up with the first version of periodic table in 1864. Mendeleev's table was based on the atomic weight of the elements. He found that there were many elements that shared similar properties and occurred periodically. Thus, he came up with the concept of periodic table of elements. This table had been divided into three main sections - metals, nonmetals, and metalloids. In this article, we shall concentrate on the metalloids from the periodic table. According to Mendeleev's law of periodic table, the chemical and physical properties of elements vary in a periodic fashion according to their atomic weights. 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When boron reacts with sodium, it acts as a non-metal, whereas in case of reaction with fluorine, boron exhibits metallic properties. History of the Periodic TableThe periodic table helps us to classify and compare various elements on the basis of their chemical behavior. Read on to know how the periodic table evolved over a period... Periodic Table with Atomic MassThe periodic table provides us with a comprehensive view of the various elements. Along with atomic mass and atomic number values, this table helps us to understand the various properties... Home Science Health and Arthritis: History Literature and Language Technology Health [25 Law & Legal Issues Business & Finance All Topics Random Leaderboard Related Topics Boron Chemistry Periodic Table Elements and Compounds in order to continue enjoying our site, we ask that you confirm your identity as a human. Thank you very much for your cooperation. [Alkali Metals] [Alkaline Earth Metals] [Halogens] [Lanthanides/Actinides] [Liquids (top)] [Main Group] [Metalloids] [Metals] [Noble Gases] [Non-Metals] [Solids (top)] [Transition Metals] [Periodic Table] The metalloids: boron (B), silicon (Si), germanium (Ge), arsenic (As), antimony (Sb), tellurium (Te), polonium (Po), astatine (At) are the elements found along the step line between metals and non-metals of the periodic table. Metalloids have properties of both metals and non-metals. Some of the metalloids, such as silicon and germanium, are useful in semi-conductors. This property makes metalloids useful in electronic components. Some allotropes of elements show more pronounced metal, metalloid or non-metal behavior than others. The element carbon, its diamond allotrope is non-metallic, however the graphite allotrope is electrically conductive showing characteristic more like a metalloid. Phosphorus, tin, selenium and bismuth also have allotropes which display borderline behavior. Point at or click on element for more information. 1A1A2 1A2A33H3B43VB4H63VB6H73VB7H83VB8H93VB9H103VB10H113VB11H123VB12H133VA13VA23VA33VA43VA53VA63VA73VA83VA93VA103VA113VA123VA133VA143VA153VA163VA173VA183VA193VA203VA213VA223VA233VA243VA253VA263VA273VA283VA293VA303VA313VA323VA333VA343VA353VA363VA373VA383VA393VA403VA413VA423VA433VA443VA453VA463VA473VA483VA493VA503VA513VA523VA533VA543VA553VA563VA573VA583VA593VA603VA613VA623VA633VA643VA653VA663VA673VA683VA693VA703VA713VA723VA733VA743VA753VA763VA773VA783VA793VA803VA813VA823VA833VA843VA853VA863VA873VA883VA893VA903VA913VA923VA933VA943VA953VA963VA973VA983VA993100
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